

#1 Dye laser

The molecule acts as a box of 4.18 nm where the electron is confined.

a) The wave functions are what we expect from a standing wave between two closed boundaries.

mode 1  $\lambda_1 = 2L$ $k_1 = \frac{2\pi}{\lambda_1} = \frac{\pi}{L}$

mode 2  $\lambda_2 = L$ $k_2 = \frac{2\pi}{L}$ with $L = 4.18 \text{ nm}$

mode 3  $\lambda_3 = \frac{2L}{3}$ $k_3 = \frac{3\pi}{L}$

From the wave vectors, we can calculate the energies,

$$E_1 = \frac{\hbar^2 k_1^2}{2m} = \frac{(1.054 \cdot 10^{-34} \text{ J}\cdot\text{s})^2 \cdot \left(\frac{\pi}{4.18 \cdot 10^{-9} \text{ m}}\right)^2}{2 \cdot 9.11 \cdot 10^{-31} \text{ kg}} = 3.45 \text{ J} = 0.0215 \text{ eV}$$

replacing k_1 by k_2 and k_3 , we get

$E_1 = 0.0215 \text{ eV}$ $E_2 = 0.0861 \text{ eV}$ $E_3 = 0.194 \text{ eV}$
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#1 b) ^{2/14} The energy of the emitted photon is equal to the difference in energy between the initial and final states of the electron.

$$E_{\lambda_{n=2 \rightarrow n=1}} = E_2 - E_1 = \underline{0.065 \text{ eV}}$$

This photon is about 50 times less energetic than a photon in the visible range. It corresponds to an Infrared photon.

c) We want to find the energy of the second transition:

$$E_{\lambda_{n=3 \rightarrow n=1}} = E_3 - E_1 = \underline{0.172 \text{ eV}}$$

where E_1 is the ground state energy,
i.e. the lowest energy.

*2 Uncertainty principle for a particle in a box.

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The box has length L , and the wave function in the ground state is $\Psi_1(x) = A \sin(k_1 x) = \frac{A}{2i} (e^{ik_1 x} - e^{-ik_1 x})$

(to get the other standing states, we just have to)
replace k_1 by k_n

a) The different k_n must satisfy the fixed boundary condition in $x=L$.

$$\rightarrow \sin(k_n L) = 0 \Rightarrow k_n L = n\pi \Rightarrow \boxed{k_n = \frac{n\pi}{L}}$$

b) In the ground state ($n=1$), the wave function is

$$\Psi_1(x) = \frac{A}{2i} (e^{ik_1 x} - e^{-ik_1 x})$$

using the relation $|p| = \hbar |k|$, we can see that the wave function correspond to the sum of one travelling wave with momentum $p_+ = \hbar k$, and another with $p_- = -\hbar k$,

As for their probability, since probability is proportional to the square of the amplitude and since

#2 b) continued

both travelling waves have the same coefficient in Ψ ,
their probability will be equal!

So $p_{1+} = \hbar k_1$, $p_{1-} = -\hbar k_1$ in x-direction with equal probability.

$$k_1 = \frac{\pi}{L}$$

c) Hence we don't know if we have p_{1+} or p_{1-} .

Then the uncertainty is $\Delta p = \frac{(p_{1+} - p_{1-})}{2} = \frac{2\pi\hbar}{2L} (= \hbar k_1)$

$$\Delta p_x = \frac{\pi\hbar}{L}$$

d) If $\Delta x = \frac{L}{2}$, then $\rightarrow$ using $\frac{x_{\max} - x_{\min}}{2} = \frac{L}{2}$

$$\Delta x \Delta p = \frac{L}{2} \cdot \frac{\pi\hbar}{L} = \frac{\pi\hbar}{2}$$

We hence get a relation satisfying to uncertainty principle.

The extra factor of $\frac{\pi}{2}$ is small and we are close to the limit of the uncertainty principle.

#3 Size of the Hydrogen atom

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- a) To achieve this, we can use the Heisenberg Uncertainty Principle. We assume that the electron is confined in a box of length $2r$. This r will correspond to the radius where it is most probable to find the electron.

We have $\Delta x = r$. So $\Delta x \Delta p_x = \hbar \rightarrow \boxed{\Delta p_x = \frac{\hbar}{r}}$

- b) Hence $\Delta p = \frac{\hbar}{r}$. Since the average ^{momentum} ~~displacement~~ of the electron is 0, this means that $\frac{1}{2}(p_{\max} - p_{\min}) = \Delta p$ and $p_{\max} + p_{\min} = 0$. So $p = \pm \frac{\hbar}{2r}$ from what we have seen in #2.

• From this $KE = \frac{p^2}{2m} = \frac{\hbar^2}{2mr^2}$ so $\boxed{KE = \frac{\hbar^2}{2mr^2}}$

- c) • The potential energy is due to Coulomb interaction between the electron and the hydrogen nucleus which are of charge $-e$ and $+e$ respectively.

#3 continued

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In general the potential energy is given by

$$PE = \frac{kq_1q_2}{r} \quad \text{where } k = \frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \frac{\text{N} \cdot \text{m}^2}{\text{C}^2}$$

$$\text{so } PE = -\frac{ke^2}{r}$$

d). To find the r minimizing the total energy
(this will be the most probable r , let call this a_0)

$$\text{so } \left. \frac{dE}{dr} \right|_{r=a_0} = 0 = \left(\frac{dKE}{dr} + \frac{dPE}{dr} \right) = \left(\frac{ke^2}{r^2} - \frac{\hbar^2}{mr^3} \right)_{r=a_0}$$

$$\text{so } \boxed{a_0 = \frac{\hbar^2}{mke^2} = 0.529 \text{ \AA}}$$
 This value is usually noted a_0 and called the Bohr radius.

A diameter of $\sim 1 \text{ \AA}$ for the hydrogen atom is very reasonable, even if ~~we~~ use the uncertainty principle, we get an excellent result here.

e). so the energy at $r=a_0$ is $E_0 = \frac{-ke^2}{a_0} + \frac{\hbar^2}{2ma_0^2}$

$$\text{and } E_0 = -9 \times 10^9 \frac{\text{J} \cdot \text{m}}{\text{C}^2} \frac{(1.6 \times 10^{-19} \text{ C})^2}{0.529 \times 10^{-10} \text{ m}} + \frac{(1.054 \cdot 10^{-34} \text{ J} \cdot \text{s})^2}{2 \cdot 9.11 \times 10^{-31} \text{ kg} (0.529 \times 10^{-10} \text{ m})^2}$$

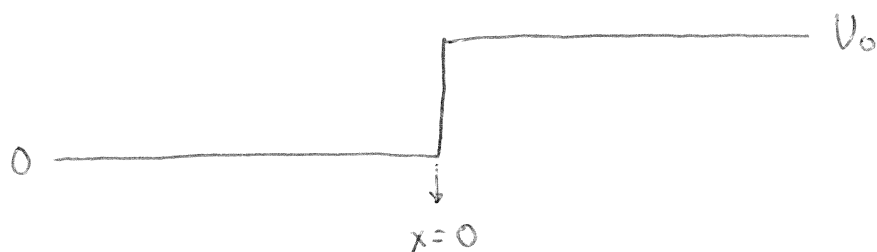
$$= (-4.36 + 2.18) e^{-18} \text{ J} = -27.22 + 13.62 \text{ eV} = -13.6 \text{ eV}$$

so $\boxed{E_0 \approx E_{\text{ionization}} = -13.6 \text{ eV}}$ it represents the binding energy of the electron to the nucleus.

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Potential energy:

and $U_0 \in]-\infty, +\infty[$

a) We know that $KE = \frac{p^2}{2m} \rightarrow p = \sqrt{2mKE}$.

there are two cases, the incoming particle correspond to the first

• $x \leq 0$, $KE = E \rightarrow \boxed{p_{x \leq 0} = p_1 = \sqrt{2mE}}$ momentum
of incoming
particle

• $x \geq 0$, $KE = E - PE = E - U_0 \rightarrow \boxed{p_{x \geq 0} = p_2 = \sqrt{2m(E - U_0)}}$

the last value is a real number since $E > U_0$

b) r is the reflected amplitude of wave function.

It will give the probability of having a reflected particle at the boundary.

t is the amplitude of wave function for transmitted particle.

* 4 c) Since $k = \frac{p}{\hbar}$

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$$k_1 = \frac{\sqrt{2mE}}{\hbar}$$

$$k_2 = \frac{\sqrt{2m(E-U_0)}}{\hbar} \rightarrow \text{see part a)}$$

d) $x \leq 0$ $\frac{d^2}{dx^2} (e^{ik_1 x} + \underline{r} e^{-ik_1 x}) = -k_1^2 (e^{ik_1 x} + \underline{r} e^{-ik_1 x})$

so $-\frac{\hbar^2}{2m} \frac{d^2}{dx^2} \psi_1(x) + U(x) \psi_1(x) = \frac{\hbar^2 k_1^2}{2m} \psi_1(x) + 0 \cdot \psi_1(x)$

and $\frac{\hbar^2 k_1^2}{2m} = E = E \psi_1(x) \quad \checkmark$

$x \geq 0$ $-\frac{\hbar^2}{2m} \frac{d^2}{dx^2} \psi_2(x) + U(x) \psi_2(x) = \frac{\hbar^2 k_2^2}{2m} \psi_2(x) + U_0 \psi_2(x)$

and $\frac{\hbar^2 k_2^2}{2m} = E - U_0 = E \psi_2(x) \quad \checkmark$

e) In $x=0$, we must have $\psi_1(0) = \psi_2(0)$

so $\boxed{1 + \underline{r} = 1}$ as usual!

* 4 f) The derivatives must also be continuous:

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$$\left. \frac{d}{dx} \psi_1(x) \right|_{x=0} = \left. \frac{d}{dx} \psi_2(x) \right|_{x=0}$$

$$ik_1 (e^{ik_1 x} - \underline{r} e^{ik_1 x}) \Big|_{x=0} = ik_2 \underline{t} e^{ik_2 x} \Big|_{x=0}$$

$$\boxed{k_1 (1 - \underline{r}) = k_2 \underline{t}}$$

g) So $1 - \underline{r} = \frac{k_2}{k_1} \underline{t}$ and $1 + \underline{r} = \underline{t}$ from f) and e)

adding these, we get $2 = (1 + \frac{k_2}{k_1}) \underline{t} \Rightarrow \underline{t} = \frac{2k_1}{k_1 + k_2}$

then, $\underline{r} = \underline{t} - 1 = \frac{2k_1 - k_1 - k_2}{k_1 + k_2} \Rightarrow \underline{r} = \frac{k_1 - k_2}{k_1 + k_2}$

h) The probability of being reflected is hence

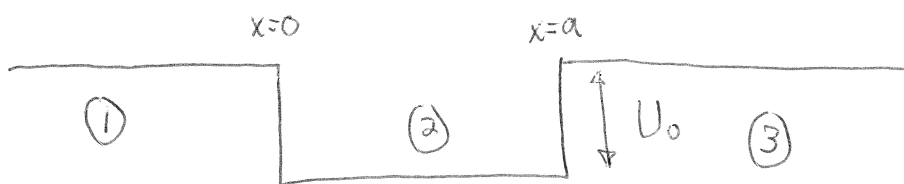
$$P(\text{reflected}) = |\underline{r}|^2 = \left(\frac{k_1 - k_2}{k_1 + k_2} \right)^2$$

Classically the probability would always

be 0 since $E > U_0$, everything should be transmitted.

#5. $E > 0$, Potential given by

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with $U_0 < 0$.

so $\psi_1(x) = e^{ikx} + \underline{R} e^{-ikx}$

$$\psi_3(x) = \underline{I} e^{ikx}$$

Want to find $\underline{R}$ and $\underline{I}$.

1st method, resolution of the Schrodinger equation

5.1 a) The wave function ψ_2 will have the form

$$\psi_2(x) = \underline{C} e^{ik'x} + \underline{D} e^{-ik'x}$$

To find k and k' , we use the relations involving kinetic energy:

$$KE = \frac{p^2}{2m} \text{ and } k = \frac{p}{\hbar} \rightarrow k = \frac{1}{\hbar} \sqrt{2m KE}$$

Since in region 1 and 3 $PE=0$, $KE=E$ and

$$k = \frac{1}{\hbar} \sqrt{2m E}$$

In region 2, $KE = E - U_0 > 0$ and

$$k' = \frac{1}{\hbar} \sqrt{2m (E - U_0)}$$

5.1 b) Both the functions and their first derivative should 11/14 be continuous at the boundaries. Hence

$$\psi_1(x=0) = \psi_2(x=0) \Rightarrow \boxed{1 + \underline{R} = \underline{C} + \underline{D}} \quad (i)$$

$$\left. \frac{d\psi_1}{dx} \right|_{x=0} = \left. \frac{d\psi_2}{dx} \right|_{x=0} \Rightarrow \boxed{k(1 - \underline{R}) = k'(\underline{C} - \underline{D})} \quad (ii)$$

c) Same thing in $x=a$

$$\psi_3(x=a) = \psi_2(x=a) \Rightarrow \boxed{\underline{I} e^{ika} = \underline{C} e^{ik'a} + \underline{D} e^{-ik'a}} \quad (iii)$$

and finally,

$$\left. \frac{d\psi_3}{dx} \right|_{x=a} = \left. \frac{d\psi_2}{dx} \right|_{x=a} \Rightarrow \boxed{k \underline{I} e^{ika} = k'(\underline{C} e^{ik'a} - \underline{D} e^{-ik'a})} \quad (iv)$$

d) We want to solve this for $\underline{R}$.

$$\begin{aligned} (1) \quad k \cdot (i) + (ii) &\Rightarrow 2k = k(\underline{C} + \underline{D}) + k'(\underline{C} - \underline{D}) \\ &= (k+k')\underline{C} + (k-k')\underline{D} \end{aligned}$$

$$\boxed{\underline{C} = \frac{(k'-k)}{k+k'} \underline{D} + \frac{2k}{k+k'}} \quad (v)$$

$$\begin{aligned} (2) \quad k \cdot (i) - (ii) &\quad 2k\underline{R} = (k-k')\underline{C} + (k+k')\underline{D} \\ &= \left(-\frac{(k-k')^2}{k+k'} + k+k' \right) \underline{D} + \frac{2k}{k+k'} (k-k') \\ &= \frac{4kk'}{k+k'} \underline{D} + 2k \frac{(k-k')}{k+k'} \end{aligned}$$

using (v)

5.1 d) continued

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so
$$\boxed{R = \frac{2k'}{k+k'} \underline{D} + \frac{(k-k')}{k+k'}} \quad \text{vi)}$$

(3) k. iii) - iv)
$$0 = \underline{C} e^{ik'a} (k-k') + \underline{D} e^{-ik'a} (k+k')$$

using v)

$$= e^{ik'a} (k-k') \cdot \left(\frac{(k'-k)}{k+k'} \underline{D} + \frac{2k}{k+k'} \right) + \underline{D} e^{ik'a} (k+k')$$

so

$$-2k \frac{(k-k')}{k+k'} e^{ik'a} = \underline{D} \left(-e^{ik'a} \frac{(k'-k)^2}{k+k'} + e^{-ik'a} (k+k') \right)$$

so
$$\boxed{\underline{D} = \frac{2k}{k+k'} \left(\frac{k-k'}{k+k'} + \frac{k+k'}{k'-k} e^{-2ik'a} \right)^{-1}} \quad \text{vii)}$$

(4) Replacing $\underline{D}$ in eq. vi), we get

$$\underline{R} = \frac{2k'}{k+k'} \cdot \frac{2k}{k+k'} \left(\frac{k-k'}{k+k'} + \frac{k+k'}{k'-k} e^{-2ik'a} \right)^{-1} + \frac{k-k'}{k+k'}$$

$$= \frac{4kk'}{(k+k')^2} \cdot \left(\frac{k+k'}{k'-k} e^{-2ik'a} \right)^{-1} \left(-\frac{(k'-k)^2}{(k+k')^2} e^{2ik'a} + 1 \right)^{-1} + \frac{k-k'}{k+k'}$$

$$\boxed{\underline{R} = \frac{\left(\frac{2k}{k+k'} \right) \left(\frac{2k'}{k+k'} \right) \left(\frac{k'-k}{k+k'} \right) e^{2ik'a}}{\left(1 - \left(\frac{k'-k}{k+k'} \right)^2 e^{2ik'a} \right)} + \frac{k-k'}{k+k'}}$$

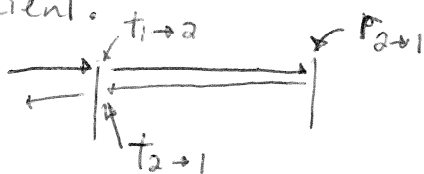
#S.2 2nd method, using the sum over history.

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a) We know that the reflexion coefficient will be $\underline{r}_{1 \rightarrow 2} = \underline{r}_{12}$, and the wave will have momentum $-hk$.

$$\text{so } \boxed{\psi_{r_a}(x) = \underline{r}_{12} e^{-ikx}}$$

b) For the wave reflected at $x=a$, we'll still have momentum $-hk$ in the end. The change will be in the coefficient.



→ also an extra phase of $2k'a$

$$\boxed{\psi_{r_b}(x) = t_{1 \rightarrow 2} t_{2 \rightarrow 1} r_{2 \rightarrow 1} e^{i2k'a} e^{-ikx}}$$

c) Summing all reflexions, the ratio will be $(r_{2 \rightarrow 1})^2 e^{i2k'a}$

$$\text{and } \sum_{i=0}^{\infty} x^i = \frac{1}{1-x}$$

We can use this for the wave in b) and then add the wave found in a).

Hence,

$$\boxed{\underline{R} = \underline{r}_{12} + \frac{t_{12} t_{21} r_{21} e^{i2k'a}}{1 - r_{21}^2 e^{i2k'a}}}$$

This is the same as what we have seen for string waves!

5.2 c) continued

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using #4.

$$r_{12} = \frac{k-k'}{k+k'}$$

$$r_{21} = \frac{k'-k}{k+k'}$$

$$t_{12} = \frac{2k}{k+k'}$$

$$t_{21} = \frac{2k'}{k+k'}$$

So

$$\underline{R} = \frac{k-k'}{k+k'} + \frac{\left(\frac{2k}{k+k'}\right)\left(\frac{2k'}{k+k'}\right)\left(\frac{k'-k}{k+k'}\right)e^{i2k'a}}{1 - \left(\frac{k'-k}{k+k'}\right)^2 e^{i2k'a}}$$

The same as in #5.1!